

Notes 02/27

Wednesday, February 27, 2008
9:56 AM

Check handouts on VOH

List of structures to know

Closed packing

Only 2 types of closest packing

Hexagonal closest packing (ABAB)

Cubic closest packing (ABC)

Other types of Packing (not closest)

Cubic packing

Body center cubic packing

- Four forms of carbon

- Graphite

- Shows electrical conductivity

- $C + Li$ or $K \rightarrow Li^+ + C^-$

- 1 K for each 8 carbons ($K^+C_8^-$)

- $C_{24}K$, $C_{36}K$, $C_{48}K$, $C_{60}K$

- ◆ K or Li intercalatoin - diffusion of K^+ or Li^+ between layers of graphite.

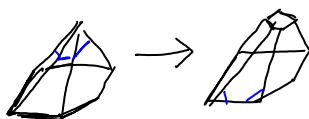
- Diamond (draw.. Will be on exam)

- Tetrahedral carbons (extended structure based on cubic closest packing of C)

- Fullerenes (will not have to draw on exam)

- C_{60}

- Icosahedron - chop of corners of icosahedron... leaves behind pentagons.

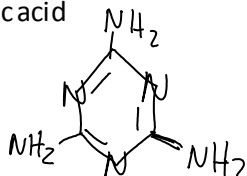


- Ferrocene



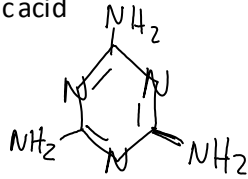
- When C is connected to electronegative atoms than it can do hydrogen bonding

- Cyanuric acid

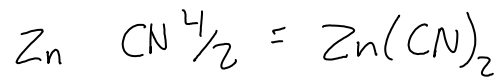


- $Zn(CN)_2$ (based on diamond)

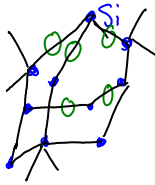
- Cyanuric acid



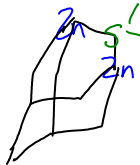
- $\text{Zn}(\text{CN})_2$ (based on diamond)



- SiO_2

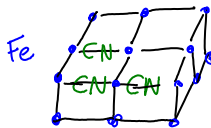


- ZnS



Alternating Zn and S

- $\text{FeFe}(\text{CN})_6$

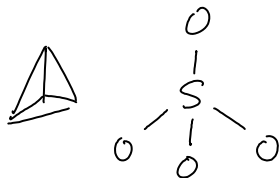


- $\text{KFeFe}(\text{CN})_6$

- Mixed iron (III) and iron (II)

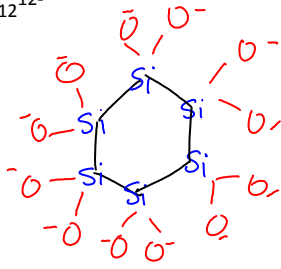
- Silicon

- Much of silicon is silicon oxides (rock)



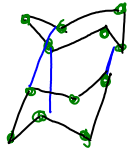
Know other structures on VOH

- $\text{Si}_6\text{O}_{12}^{12-}$



- SiO_2

Hexagonal packing



Cubic packing (see VOH)